Chemistry

UC Berkeley offers two bachelors’ degrees in Chemistry: a Bachelor of Science (BS) through the College of Chemistry and a Bachelor of Arts (BA) through the College of Letters and Science. For specific information regarding degree requirements for each, please refer to the information below, and the appropriate Major Requirements and College Requirements tabs on this page.

BS in Chemistry, College of Chemistry

The BS in Chemistry is intended for students who are primarily interested in careers as professional chemists or wish to have a thorough grounding in chemistry in preparation for professional or graduate school in chemistry and related disciplines. Students in the BS program may also choose to pursue the Materials Chemistry concentration.

Students interested in subsequent graduate studies in Chemistry will receive a better preparation by pursuing the BS in Chemistry.

BA in Chemistry, College of Letters & Science

The BA in Chemistry includes a greater number of humanities and social science courses than the Bachelor of Science degree and is intended for those interested in careers in teaching, medicine, or other sciences in which a basic understanding of chemical processes is necessary.

Students who want to pursue the BA degree should apply for admission to the College of Letters & Science.

Minor Program

The College of Chemistry offers a minor in Chemistry. Chemical biology majors are not eligible to pursue this minor. Students must submit a notification of completion of the minor to the College of Chemistry Undergraduate Advising Office.

In addition to the University, campus, and college requirements, listed on the College Requirements tab, students must fulfill the below requirements specific to their major program.

General Guidelines

1. A minimum grade point average (GPA) of 2.0 must be maintained in all courses undertaken at UC Berkeley, including those from UC Summer Sessions, UC Education Abroad Program, UC Berkeley in Washington Program, and XB courses from University Extension.

2. A minimum GPA of 2.0 in all courses taken in the college is required in order to advance and continue in upper division courses.

3. A minimum GPA of 2.0 in all upper division courses taken at the University is required to satisfy major requirements.

4. Chemistry majors who receive a grade of D+ or lower in a chemistry course for which a grade of C- or higher is required must repeat the course at UC Berkeley.

For information regarding grade requirements in specific courses, please see the notes sections below.

For information regarding residence requirements and unit requirements, please see the College Requirements tab.

Lower Division Requirements

<table>
<thead>
<tr>
<th>Course</th>
<th>Title</th>
<th>Units</th>
</tr>
</thead>
<tbody>
<tr>
<td>CHEM 4A</td>
<td>General Chemistry and Quantitative Analysis</td>
<td>4</td>
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<tr>
<td>CHEM 4B</td>
<td>General Chemistry and Quantitative Analysis</td>
<td>4</td>
</tr>
<tr>
<td>CHEM 12A</td>
<td>Organic Chemistry (Formerly 112A)</td>
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<tr>
<td>CHEM 12B</td>
<td>Organic Chemistry</td>
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<tr>
<td>MATH 1A</td>
<td>Calculus</td>
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<tr>
<td>MATH 1B</td>
<td>Calculus</td>
<td>4</td>
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<tr>
<td>MATH 53</td>
<td>Multivariable Calculus</td>
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<tr>
<td>MATH 54</td>
<td>Linear Algebra and Differential Equations</td>
<td>4</td>
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<tr>
<td>PHYSICS 7A</td>
<td>Physics for Scientists and Engineers</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 7B</td>
<td>Physics for Scientists and Engineers</td>
<td>4</td>
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</tbody>
</table>

Notes

1. Students should take CHEM 4A and CHEM 4B during their freshman year, and CHEM 12A and CHEM 12B during their sophomore year.

2. A grade of C- or better is required in CHEM 4A before taking CHEM 4B, in CHEM 4B before taking more advanced courses, and in CHEM 12A before taking CHEM 12B.

3. A grade of C- or better is recommended in CHEM 12A before taking BIOLOGY 1A.

4. Students who join the program after completing a general chemistry sequence that does not include quantitative analysis are required to take CHEM 4B, CHEM 15, or CHEM 105.

5. Students who join the program after completing CHEM 3A plus CHEM 3AL and CHEM 3BL at Berkeley are allowed to substitute those courses for CHEM 12A and CHEM 12B.

6. Students who join the program after completing only CHEM 3A plus CHEM 3AL at Berkeley are recommended to take CHEM 112B.

7. Students should start MATH 1A in the first semester of their freshman year.

7. Students should start PHYSICS 7A in the second semester of their freshman year.

Upper Division Requirements

For information regarding the upper division requirements for the Materials Chemistry concentration, see below.

<table>
<thead>
<tr>
<th>Course</th>
<th>Title</th>
<th>Units</th>
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<tbody>
<tr>
<td>CHEM 104A</td>
<td>Advanced Inorganic Chemistry</td>
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<tr>
<td>CHEM 104B</td>
<td>Advanced Inorganic Chemistry</td>
<td>3</td>
</tr>
<tr>
<td>CHEM 120A</td>
<td>Physical Chemistry</td>
<td>3</td>
</tr>
<tr>
<td>CHEM 120B</td>
<td>Physical Chemistry</td>
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<tr>
<td>CHEM 125</td>
<td>Physical Chemistry Laboratory ¹</td>
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<tr>
<td>Select one of the following:</td>
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<tr>
<td>CHEM 105</td>
<td>Instrumental Methods in Analytical Chemistry</td>
<td></td>
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<tr>
<td>CHEM 108</td>
<td>Inorganic Synthesis and Reactions</td>
<td></td>
</tr>
<tr>
<td>CHEM 115</td>
<td>Organic Chemistry--Advanced Laboratory Methods</td>
<td></td>
</tr>
<tr>
<td>CHEM 146</td>
<td>Radiochemical Methods in Nuclear Technology and Forensics</td>
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</table>

Select 12 units of upper division chemistry and allied subjects courses (see below) ²

One course must be an additional lecture course (or lab/lecture course) as approved by the student's staff adviser.

¹ CHEM C182 may be substituted for CHEM 125
Allied Subjects Courses

<table>
<thead>
<tr>
<th>Course Code</th>
<th>Course Title</th>
<th>Units</th>
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</thead>
<tbody>
<tr>
<td>CHM ENG 195</td>
<td>The Berkeley Lectures on Energy: Energy from Biomass</td>
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<tr>
<td>CHM ENG 196</td>
<td>Special Laboratory Study</td>
<td>2-4</td>
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<tr>
<td>CHEM 100</td>
<td>Communicating Chemistry (limited to 2 units)</td>
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<tr>
<td>CHEM 103</td>
<td>Inorganic Chemistry in Living Systems</td>
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<tr>
<td>CHEM 104A</td>
<td>Advanced Inorganic Chemistry</td>
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<td>CHEM 104B</td>
<td>Advanced Inorganic Chemistry</td>
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<td>CHEM 105</td>
<td>Instrumental Methods in Analytical Chemistry</td>
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<tr>
<td>CHEM 108</td>
<td>Inorganic Synthesis and Reactions</td>
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<tr>
<td>CHEM C110L</td>
<td>General Biochemistry and Molecular Biology Laboratory</td>
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<tr>
<td>CHEM 113</td>
<td>Advanced Mechanistic Organic Chemistry</td>
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<tr>
<td>CHEM 114</td>
<td>Advanced Synthetic Organic Chemistry</td>
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<td>CHEM 115</td>
<td>Organic Chemistry--Advanced Laboratory Methods</td>
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<tr>
<td>CHEM 122</td>
<td>Quantum Mechanics and Spectroscopy</td>
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<tr>
<td>CHEM 125</td>
<td>Physical Chemistry Laboratory</td>
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<tr>
<td>CHEM C130</td>
<td>Biophysical Chemistry: Physical Principles and the Molecules of Life</td>
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<tr>
<td>CHEM 130B</td>
<td>Biophysical Chemistry (2 units)</td>
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<td>CHEM 135</td>
<td>Chemical Biology</td>
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<tr>
<td>CHEM C138</td>
<td>The Berkeley Lectures on Energy: Energy from Biomass</td>
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<td>CHEM 143</td>
<td>Nuclear Chemistry</td>
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<td>CHEM 146</td>
<td>Radiochemical Methods in Nuclear Technology and Forensics</td>
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<td>CHEM C150</td>
<td>Introduction to Materials Chemistry</td>
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<td>CHEM C170L</td>
<td>Biochemical Engineering Laboratory</td>
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<td>CHEM C178</td>
<td>Polymer Science and Technology</td>
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<td>CHEM C182</td>
<td>Atmospheric Chemistry and Physics Laboratory</td>
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<td>CHEM C191</td>
<td>Quantum Information Science and Technology</td>
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<td>CHEM 192</td>
<td>Individual Study for Advanced Undergraduates</td>
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<td>CHEM H194</td>
<td>Research for Advanced Undergraduates</td>
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<td>CHEM 195</td>
<td>Special Topics</td>
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<td>Air Pollution</td>
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<td>CIV ENG C116</td>
<td>Chemistry of Soils</td>
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<td>Engineering Analysis Using the Finite Element Method</td>
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<td>CIV ENG 165</td>
<td>Concrete Materials, Construction, and Sustainability</td>
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<td>User Interface Design and Development</td>
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<td>Operating Systems and System Programming</td>
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<td>Programming Languages and Compilers</td>
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<td>COMPSCI 170</td>
<td>Efficient Algorithms and Intractable Problems</td>
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<td>COMPSCI 174</td>
<td>Combinatorics and Discrete Probability</td>
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<td>COMPSCI 184</td>
<td>Foundations of Computer Graphics</td>
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<td>Quantum Information Science and Technology</td>
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<td>EPS 103</td>
<td>Introduction to Aquatic and Marine Geochemistry</td>
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<td>Petroleum Geology</td>
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<td>EPS C162</td>
<td>Planetary Astrophysics</td>
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<td>EPS C181</td>
<td>Atmospheric Physics and Dynamics</td>
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<td>EPS C182</td>
<td>Atmospheric Chemistry and Physics Laboratory</td>
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<td>EPS C183</td>
<td>Carbon Cycle Dynamics</td>
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<td>ECON C103</td>
<td>Introduction to Mathematical Economics</td>
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<td>Special Problems in Mathematics, Science and Technology Education (graduate-level; requires consent of instructor)</td>
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<td>Mathematical Thinking and Problem Solving (graduate-level; requires consent of instructor)</td>
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<td>Methods of Engineering Analysis</td>
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<td>ENGIN 128</td>
<td>Advanced Engineering Design Graphics</td>
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<td>ESPM C138</td>
<td>Introduction to Comparative Virology</td>
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<td>ESPM C148</td>
<td>Pesticide Chemistry and Toxicology</td>
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<td>Probability and Risk Analysis for Engineers</td>
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<td>INTEGBI 106A</td>
<td>Physical and Chemical Environment of the Ocean</td>
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<td>Introduction to Systems in Biology and Medicine</td>
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<td>MAT SCI 102</td>
<td>Bonding, Crystallography, and Crystal Defects</td>
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<td>MAT SCI 103</td>
<td>Phase Transformations and Kinetics</td>
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<td>MAT SCI 104</td>
<td>Materials Characterization</td>
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<td>MAT SCI 111</td>
<td>Properties of Electronic Materials</td>
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<td>MAT SCI 112</td>
<td>Corrosion (Chemical Properties)</td>
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<td>MAT SCI 113</td>
<td>Mechanical Behavior of Engineering Materials</td>
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<td>MAT SCI 117</td>
<td>Properties of Dielectric and Magnetic Materials</td>
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<td>Biological Performance of Materials</td>
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<td>MAT SCI 120</td>
<td>Materials Production</td>
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<td>MAT SCI 121</td>
<td>Metals Processing</td>
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<td>MAT SCI 122</td>
<td>Ceramic Processing</td>
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<td>ELECTRONIC MATERIALS PROCESSING</td>
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<td>Experimental Materials Science and Design</td>
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<td>MAT SCI 140</td>
<td>Nanomaterials for Scientists and Engineers</td>
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<td>MATH 105</td>
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<td>Honors Linear Algebra</td>
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<td>Introduction to Number Theory</td>
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<td>MATH 121A</td>
<td>Mathematical Tools for the Physical Sciences</td>
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<td>MATH 123</td>
<td>Ordinary Differential Equations</td>
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<td>MATH 125A</td>
<td>Mathematical Logic</td>
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<td>MATH 126</td>
<td>Introduction to Partial Differential Equations</td>
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<td>MATH 128A</td>
<td>Numerical Analysis</td>
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<td>Numerical Analysis</td>
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<td>MATH 130</td>
<td>The Classical Geometries</td>
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<td>Introduction to the Theory of Sets</td>
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<td>Incompleteness and Undecidability</td>
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<td>Metric Differential Geometry</td>
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<td>Mathematical Methods for Optimization</td>
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<td>Honors Introduction to Complex Analysis</td>
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<td>MATH 189</td>
<td>Mathematical Methods in Classical and Quantum Mechanics</td>
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<td>MEC ENG 107</td>
<td>Mechanical Engineering Laboratory</td>
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<td>MEC ENG C115</td>
<td>Molecular Biomechanics and Mechanobiology of the Cell</td>
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<td>Structural Aspects of Biomaterials</td>
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<td>Introduction to Nanotechnology and Nanoscience</td>
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<td>MEC ENG C176</td>
<td>Orthopedic Biomechanics</td>
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<td>Engineering Analysis Using the Finite Element Method</td>
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<td>MCELLBI C100A</td>
<td>Biophysical Chemistry: Physical Principles and the Molecules of Life</td>
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<td>MCELLBI C103</td>
<td>Bacterial Pathogenesis</td>
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<td>MCELLBI C104</td>
<td>Genetics, Genomics, and Cell Biology</td>
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<td>MCELLBI C110</td>
<td>Molecular Biology: Macromolecular Synthesis and Cellular Function</td>
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<td>General Biochemistry and Molecular Biology Laboratory</td>
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<td>MCELLBI C112</td>
<td>General Microbiology</td>
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<td>General Microbiology Laboratory</td>
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<td>MCELLBI C114</td>
<td>Introduction to Comparative Virology</td>
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<td>MCELLBI C116</td>
<td>Microbial Diversity</td>
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<td>MCELLBI 118</td>
<td>The Cancer Karyotype: What it is and What it Does</td>
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<td>MCELLBI 133L</td>
<td>Physiology and Cell Biology Laboratory</td>
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<td>Topics in Cell and Development Biology: Molecular Endocrinology</td>
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<td>MCELLBI 140</td>
<td>General Genetics</td>
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<td>MCELLBI 143</td>
<td>Evolution of Genomes, Cells, and Development</td>
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<td>MCELLBI C148</td>
<td>Microbial Genomics and Genetics</td>
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<td>MCELLBI 150</td>
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<td>Immunology Laboratory</td>
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<td>Nuclear Reactions and Radiation</td>
<td>4</td>
</tr>
<tr>
<td>NUC ENG 104</td>
<td>Radiation Detection and Nuclear Instrumentation Laboratory</td>
<td>4</td>
</tr>
<tr>
<td>NUC ENG 107</td>
<td>Introduction to Imaging</td>
<td>3</td>
</tr>
<tr>
<td>NUC ENG 120</td>
<td>Nuclear Materials</td>
<td>4</td>
</tr>
<tr>
<td>NUC ENG 124</td>
<td>Radioactive Waste Management</td>
<td>3</td>
</tr>
<tr>
<td>Course Code</td>
<td>Course Name</td>
<td>Units</td>
</tr>
<tr>
<td>------------</td>
<td>------------------------------------------------------</td>
<td>-------</td>
</tr>
<tr>
<td>NUC ENG 130</td>
<td>Analytical Methods for Non-proliferation</td>
<td>3</td>
</tr>
<tr>
<td>NUC ENG 150</td>
<td>Introduction to Nuclear Reactor Theory</td>
<td>4</td>
</tr>
<tr>
<td>NUC ENG 161</td>
<td>Nuclear Power Engineering</td>
<td>4</td>
</tr>
<tr>
<td>NUC ENG 162</td>
<td>Radiation Biophysics and Dosimetry</td>
<td>3</td>
</tr>
<tr>
<td>NUC ENG 170A</td>
<td>Nuclear Design: Design in Nuclear Power Technology and Instrumentation</td>
<td>3</td>
</tr>
<tr>
<td>NUC ENG 170B</td>
<td>Nuclear Design: Design in Bionuclear, Nuclear Medicine, and Radiation Therapy</td>
<td>3</td>
</tr>
<tr>
<td>NUSCTX 103</td>
<td>Nutrient Function and Metabolism</td>
<td>3</td>
</tr>
<tr>
<td>NUSCTX 108A</td>
<td>Introduction and Application of Food Science</td>
<td>3</td>
</tr>
<tr>
<td>NUSCTX 110</td>
<td>Toxicology</td>
<td>4</td>
</tr>
<tr>
<td>NUSCTX 115</td>
<td>Principles of Drug Action</td>
<td>2</td>
</tr>
<tr>
<td>NUSCTX 160</td>
<td>Metabolic Bases of Human Health and Diseases</td>
<td>4</td>
</tr>
<tr>
<td>NUSCTX 171</td>
<td>Nutrition and Toxicology Laboratory</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 7C</td>
<td>Physics for Scientists and Engineers (must be completed with a grade of C- or better)</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 105</td>
<td>Analytic Mechanics</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 110A</td>
<td>Electromagnetism and Optics</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 110B</td>
<td>Electromagnetism and Optics</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 112</td>
<td>Introduction to Statistical and Thermal Physics</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 130</td>
<td>Quantum and Nonlinear Optics</td>
<td>3</td>
</tr>
<tr>
<td>PHYSICS 137B</td>
<td>Quantum Mechanics</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 138</td>
<td>Modern Atomic Physics</td>
<td>3</td>
</tr>
<tr>
<td>PHYSICS 141A</td>
<td>Solid State Physics</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 141B</td>
<td>Solid State Physics</td>
<td>3</td>
</tr>
<tr>
<td>PHYSICS C191</td>
<td>Quantum Information Science and Technology</td>
<td>3</td>
</tr>
<tr>
<td>PLANTBI C103</td>
<td>Bacterial Pathogenesis</td>
<td>3</td>
</tr>
<tr>
<td>PLANTBI C112</td>
<td>General Microbiology</td>
<td>4</td>
</tr>
<tr>
<td>PLANTBI C112L</td>
<td>General Microbiology Laboratory</td>
<td>2</td>
</tr>
<tr>
<td>PLANTBI C114</td>
<td>Introduction to Comparative Virology</td>
<td>4</td>
</tr>
<tr>
<td>PLANTBI C116</td>
<td>Microbial Diversity</td>
<td>3</td>
</tr>
<tr>
<td>PLANTBI 120</td>
<td>Biology of Algae</td>
<td>2</td>
</tr>
<tr>
<td>PLANTBI 120L</td>
<td>Laboratory for Biology of Algae</td>
<td>2</td>
</tr>
<tr>
<td>PLANTBI 122</td>
<td>Bioenergy</td>
<td>2</td>
</tr>
<tr>
<td>PLANTBI C124</td>
<td>The Berkeley Lectures on Energy: Energy from Biomass</td>
<td>3</td>
</tr>
<tr>
<td>PLANTBI 135</td>
<td>Physiology and Biochemistry of Plants</td>
<td>3</td>
</tr>
<tr>
<td>PLANTBI C148</td>
<td>Microbial Genomics and Genetics</td>
<td>4</td>
</tr>
<tr>
<td>PLANTBI 150</td>
<td>Plant Cell Biology</td>
<td>3</td>
</tr>
<tr>
<td>PLANTBI 160</td>
<td>Plant Molecular Genetics</td>
<td>3</td>
</tr>
<tr>
<td>PLANTBI 170</td>
<td>Modern Applications of Plant Biotechnology</td>
<td>2</td>
</tr>
<tr>
<td>PLANTBI 180</td>
<td>Environmental Plant Biology</td>
<td>2</td>
</tr>
<tr>
<td>PB HLTH C102</td>
<td>Bacterial Pathogenesis</td>
<td>3</td>
</tr>
<tr>
<td>PB HLTH 142</td>
<td>Introduction to Probability and Statistics in Biology and Public Health</td>
<td>4</td>
</tr>
<tr>
<td>PB HLTH 162A</td>
<td>Public Health Microbiology</td>
<td>3</td>
</tr>
<tr>
<td>PB HLTH 162L</td>
<td>Public Health Microbiology Laboratory</td>
<td>1</td>
</tr>
<tr>
<td>STAT 134</td>
<td>Concepts of Probability</td>
<td>4</td>
</tr>
<tr>
<td>STAT 135</td>
<td>Concepts of Statistics</td>
<td>4</td>
</tr>
<tr>
<td>STAT 140</td>
<td>Probability for Data Science</td>
<td>4</td>
</tr>
</tbody>
</table>

1. Chemistry majors can only count 2 of the 3 units towards their Allied Subject requirement.

### Upper Division Requirements: Materials Chemistry Concentration

<table>
<thead>
<tr>
<th>Course Code</th>
<th>Course Name</th>
<th>Units</th>
</tr>
</thead>
<tbody>
<tr>
<td>CHEM 104A</td>
<td>Advanced Inorganic Chemistry</td>
<td>3</td>
</tr>
<tr>
<td>CHEM 104B</td>
<td>Advanced Inorganic Chemistry</td>
<td>3</td>
</tr>
<tr>
<td>CHEM 120A</td>
<td>Physical Chemistry</td>
<td>3</td>
</tr>
<tr>
<td>CHEM 120B</td>
<td>Physical Chemistry</td>
<td>3</td>
</tr>
<tr>
<td>CHEM C150</td>
<td>Introduction to Materials Chemistry</td>
<td>3</td>
</tr>
</tbody>
</table>

Select two laboratory courses from the following:

- CHEM 105 | Instrumental Methods in Analytical Chemistry  
- or CHEM 125 | Physical Chemistry Laboratory  
- CHEM 108 | Inorganic Synthesis and Reactions  
- or CHEM 115 | Organic Chemistry—Advanced Laboratory Methods  

Electives. Select 10 units of the following:

- BIO ENG C118 Biological Performance of Materials  
- CHEM C178 | Polymer Science and Technology  
- MAT SCI 104 | Materials Characterization  
- MEC ENG 118 Introduction to Nanotechnology and Nanoscience  
- PHYSICS 141A Solid State Physics  
- PHYSICS 141B Solid State Physics

In addition to the University, campus, and college requirements, listed on the College Requirements tab, students must fulfill the below requirements specific to their major program.

### General Guidelines

1. All courses taken to fulfill the major requirements below must be taken for graded credit, other than courses listed which are offered on a Pass/Fail basis only. Other exceptions to this requirement are noted as applicable.
2. No more than one upper division course may be used to simultaneously fulfill requirements for a student's major and minor programs, with the exception of minors offered outside of the College of Letters & Science.
3. A minimum grade point average (GPA) of 2.0 must be maintained in both upper and lower division courses used to fulfill the major requirements.

For information regarding residence requirements and unit requirements, please see the College Requirements tab.

### Lower Division Requirements

<table>
<thead>
<tr>
<th>Course Code</th>
<th>Course Name</th>
<th>Units</th>
</tr>
</thead>
<tbody>
<tr>
<td>CHEM 4A</td>
<td>General Chemistry and Quantitative Analysis</td>
<td>4</td>
</tr>
<tr>
<td>CHEM 4B</td>
<td>General Chemistry and Quantitative Analysis</td>
<td>4</td>
</tr>
<tr>
<td>CHEM 12A</td>
<td>Organic Chemistry</td>
<td>3</td>
</tr>
<tr>
<td>CHEM 12B</td>
<td>Organic Chemistry</td>
<td>5</td>
</tr>
<tr>
<td>MATH 1A</td>
<td>Calculus</td>
<td>4</td>
</tr>
<tr>
<td>MATH 1B</td>
<td>Calculus</td>
<td>4</td>
</tr>
<tr>
<td>MATH 53</td>
<td>Multivariable Calculus</td>
<td>4</td>
</tr>
<tr>
<td>MATH 54</td>
<td>Linear Algebra and Differential Equations</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 7A</td>
<td>Physics for Scientists and Engineers</td>
<td>4</td>
</tr>
<tr>
<td>PHYSICS 7B</td>
<td>Physics for Scientists and Engineers</td>
<td>4</td>
</tr>
</tbody>
</table>
Upper Division Requirements

**CHEM 104A** Advanced Inorganic Chemistry ¹ ³
**CHEM 104B** Advanced Inorganic Chemistry ¹ ³
**CHEM 120A** Physical Chemistry ³ ³
**CHEM 120B** Physical Chemistry ³ ³

Select one of the following:

- **CHEM 105** Instrumental Methods in Analytical Chemistry
- **CHEM 108** Inorganic Synthesis and Reactions
- **CHEM 115** Organic Chemistry--Advanced Laboratory Methods
- **CHEM 125** Physical Chemistry Laboratory ³
- **CHEM C170L** Biochemical Engineering Laboratory
- **CHEM C182** Atmospheric Chemistry and Physics Laboratory

¹ CHEM 103 and CHEM 135 may be substituted for CHEM 104A and CHEM 104B.
² A grade of C- or higher is required in CHEM 120A and CHEM 120B if taken before CHEM 125.

Students who have a strong interest in an area of study outside their major often decide to complete a minor program. These programs have set requirements and are noted officially on the transcript in the memoranda section, but are not noted on diplomas.

**General Guidelines**

1. All courses taken to fulfill the minor requirements below must be taken for graded credit.
2. A minimum of three of the upper division courses taken to fulfill the minor requirements must be completed at UC Berkeley.
3. A minimum grade point average (GPA) of 2.0 is required for courses used to fulfill the minor requirements.
4. Students must consult with their college/school for information regarding overlap of courses between their majors and minors.

**Requirements**

1. Two semesters of organic chemistry (Chem 3A/L & 3BL or Chem 12A & 12B)
2. Two semesters of physical or biophysical chemistry (Chem 120A & 120B or Chem C130 & 130B)
3. Two additional upper division Chemistry courses taken at Berkeley, excluding courses numbered 190-199

**Organic chemistry options:**

<table>
<thead>
<tr>
<th>Course Code</th>
<th>Course Title</th>
</tr>
</thead>
<tbody>
<tr>
<td>CHEM 3A &amp; 3AL</td>
<td>Chemical Structure and Reactivity and Organic Chemistry Laboratory &amp; CHEM 3B &amp; 3BL</td>
</tr>
<tr>
<td>CHEM 12A &amp; 12B</td>
<td>Organic Chemistry &amp; Organic Chemistry</td>
</tr>
</tbody>
</table>

**Physical or biophysical chemistry options:**

<table>
<thead>
<tr>
<th>Course Code</th>
<th>Course Title</th>
</tr>
</thead>
<tbody>
<tr>
<td>CHEM 120A</td>
<td>Physical Chemistry &amp; CHEM 120B and Physical Chemistry</td>
</tr>
<tr>
<td>CHEM C130</td>
<td>Biophysical Chemistry: Physical Principles and the &amp; CHEM 130B Molecules of Life and Biophysical Chemistry</td>
</tr>
</tbody>
</table>

**Humanities and Social Sciences Breadth Requirement**

The College of Chemistry’s humanities and social sciences breadth requirement promotes educational experiences that enrich and complement the technical requirements for each major.

**Chemistry & Chemical Biology majors**

- 15 units total; includes Reading & Composition and American Cultures courses
- Remaining units must come from the following lists of approved humanities and social science courses, excluding courses which only teach a skill (such as drawing or playing an instrument):
  - Arts and Literature (http://guide.berkeley.edu/undergraduate/colleges-schools/chemistry/arts-literature)
  - Foreign Language (http://guide.berkeley.edu/undergraduate/colleges-schools/chemistry/approved-foreign-language-courses)
  - Historical Studies (http://guide.berkeley.edu/undergraduate/colleges-schools/chemistry/historical-studies)
  - International Studies (http://guide.berkeley.edu/undergraduate/colleges-schools/chemistry/international-studies)
  - Philosophy and Values (http://guide.berkeley.edu/undergraduate/colleges-schools/chemistry/philosophy-values)
  - Social and Behavioral Sciences (http://guide.berkeley.edu/undergraduate/colleges-schools/chemistry/social-behavioral-sciences)

- Breadth courses may be taken on a Pass/No Pass basis (excluding Reading and Composition)
• AP, IB, and GCE A-level exam credit (http://chemistry.berkeley.edu/students/current-undergraduates/exam-credit-info) may be used to satisfy the breadth requirement

1 Elementary-level courses may not be in the student’s native language and may not be structured primarily to teach the reading of scientific literature.

2 For Chemistry and Chemical Biology majors, elementary-level foreign language courses are not accepted toward the 15 unit breadth requirement if they are used (or are duplicates of high school courses used) to satisfy the Foreign Language requirement.

Foreign Language
The Foreign Language requirement may be satisfied with one foreign language, in one of the following ways:

• By completing in high school the third year of one foreign language with minimum grades of C-.
• By completing at Berkeley the second semester of a sequence of courses in one foreign language or the equivalent at another institution. Only foreign language courses that include reading and composition as well as conversation are accepted in satisfaction of this requirement. Foreign language courses may be taken on a Pass/No Pass basis.
• By demonstrating equivalent knowledge of a foreign language through examination, including a College Entrance Examination Board (CEEB) Advanced Placement Examination with a score of 3 or higher (if taken before admission to college), an SAT II: Subject Test with a score of 590 or higher, or a proficiency examination offered by some departments at Berkeley or at another campus of the University of California.

Class Schedule Requirements
• Minimum units per semester: 13
• Maximum units per semester: 19.5
• 12 units of course work each semester must satisfy degree requirements
• Chemical Engineering freshmen and Chemistry majors are required to enroll in a minimum of one chemistry course each semester
• After the freshman year, Chemical Engineering majors must enroll in a minimum of one chemical and biomolecular engineering course each semester

Semester Limit
• Students who entered as freshmen: 8 semesters
• Chemistry & Chemical Biology majors who entered as transfer students: 4 semesters
• Chemical Engineering majors who entered as transfer students: 5 semesters

Summer sessions are excluded when determining the limit on semesters. Students who wish to delay graduation to complete a minor, a double major, or simultaneous degrees must request approval for delay of graduation before what would normally be their final two semesters. The College of Chemistry does not have a rule regarding maximum units that a student can accumulate.

Senior Residence
After 90 units toward the bachelor’s degree have been completed, at least 24 of the remaining units must be completed in residence in the College of Chemistry, in at least two semesters (the semester in which the 90 units are exceeded, plus at least one additional semester).

To count as a semester of residence for this requirement, a program must include at least 4 units of successfully completed courses. A summer session can be credited as a semester in residence if this minimum unit requirement is satisfied.

Juniors and seniors who participate in the UC Education Abroad Program (UCEAP) for a full year may meet a modified senior residence requirement. After 60 units toward the bachelor’s degree have been completed, at least 24 (excluding UCEAP) of the remaining units must be completed in residence in the College of Chemistry, in at least two semesters. At least 12 of the 24 units must be completed after the student has already completed 90 units. Undergraduate dean’s approval for the modified senior residence requirement must be obtained before enrollment in the Education Abroad Program.

Minimum Total Units
A student must successfully complete at least 120 semester units in order to graduate.

Minimum Academic Requirements
A student must earn at least a C average (2.0 GPA) in all courses undertaken at UC including those from UC Summer Sessions, UC Education Abroad Program, and UC Berkeley Washington Program as well as XB courses from University Extension.

Minimum Course Grade Requirements
Students in the College of Chemistry who receive a grade of D+ or lower in a chemical and biomolecular engineering or chemistry course for which a grade of C- or higher is required must repeat the course at UC Berkeley.

Students in the College of Chemistry must achieve:

• C- or higher in CHEM 4A General Chemistry and Quantitative Analysis before taking CHEM 4B General Chemistry and Quantitative Analysis
• C- or higher in CHEM 4B General Chemistry and Quantitative Analysis before taking more advanced courses
• C- or higher in CHEM 12A Organic Chemistry before taking CHEM 12B Organic Chemistry
• GPA of at least 2.0 in all courses taken in the college in order to advance to and continue in the upper division

Chemistry or Chemical Biology majors must also achieve:

• C- or higher in CHEM 120A Physical Chemistry and CHEM 120B Physical Chemistry if taken before CHEM 125 Physical Chemistry Laboratory or CHEM C182 Atmospheric Chemistry and Physics Laboratory
• 2.0 GPA in all upper division courses taken at the University to satisfy major requirements

Chemical Engineering students must also achieve:
• C- or higher in Chemical and Biomolecular Engineering (CBE) 140 before taking any other CBE courses
• C- or higher in CHM ENG 150A Transport Processes to be eligible to take any other course in the 150 series
• 2.0 GPA in all upper division courses taken at the University to satisfy major requirements

Chemical Engineering students who do not achieve a grade of C- or higher in CHM ENG 140 Introduction to Chemical Process Analysis on their first attempt are advised to change to another major. If the course is not passed with a grade of C- or higher on the second attempt, continuation in the Chemical Engineering program is normally not allowed.

**Minimum Progress**
To make normal progress toward a degree, undergraduates must successfully complete 30 units of coursework each year. The continued enrollment of students who do not maintain normal progress will be subject to the approval of the undergraduate dean. To achieve minimum academic progress, the student must meet two criteria:

1. Completed no fewer units than 15 multiplied by the number of semesters, less one, in which the student has been enrolled at Berkeley. Summer sessions do not count as semesters for this purpose.
2. A student’s class schedule must contain at least 13 units in any term, unless otherwise authorized by the staff adviser or the undergraduate dean.

Undergraduate students in the College of Letters & Science must fulfill the following requirements in addition to those required by their major program.

For detailed lists of courses that fulfill college requirements, please review the College of Letters & Sciences page in this Guide.

**Entry Level Writing** (http://writing.berkeley.edu/node/78)
All students who will enter the University of California as freshmen must demonstrate their command of the English language by fulfilling the Entry Level Writing requirement. Fulfillment of this requirement is also a prerequisite to enrollment in all reading and composition courses at UC Berkeley.

**American History and American Institutions**
(http://guide.berkeley.edu/undergraduate/colleges-schools/letters-science/american-history-institutions-requirement)
The American History and Institutions requirements are based on the principle that a US resident graduated from an American university, should have an understanding of the history and governmental institutions of the United States.

**American Cultures**
(http://americancultures.berkeley.edu/students/courses)
American Cultures is the one requirement that all undergraduate students at Cal need to take and pass in order to graduate. The requirement offers an exciting intellectual environment centered on the study of race, ethnicity and culture of the United States. AC courses offer students opportunities to be part of research-led, highly accomplished teaching environments, grappling with the complexity of American Culture.

**Quantitative Reasoning**
The Quantitative Reasoning requirement is designed to ensure that students graduate with basic understanding and competency in math, statistics, or computer science. The requirement may be satisfied by exam or by taking an approved course.

**Foreign Language**
The Foreign Language requirement may be satisfied by demonstrating proficiency in reading comprehension, writing, and conversation in a foreign language equivalent to the second semester college level, either by passing an exam or by completing approved course work.

**Reading and Composition**
In order to provide a solid foundation in reading, writing, and critical thinking the College requires two semesters of lower division work in composition in sequence. Students must complete a first-level reading and composition course by the end of their second semester and a second-level course by the end of their fourth semester.

**Breadth Requirements**
The undergraduate breadth requirements provide Berkeley students with a rich and varied educational experience outside of their major program. As the foundation of a liberal arts education, breadth courses give students a view into the intellectual life of the University while introducing them to a multitude of perspectives and approaches to research and scholarship. Engaging students in new disciplines and with peers from other majors, the breadth experience strengthens interdisciplinary connections and context that prepares Berkeley graduates to understand and solve the complex issues of their day.

**Unit Requirements**
• 120 total units, including at least 60 L&S units
• Of the 120 units, 36 must be upper division units
• Of the 36 upper division units, 8 must be taken in courses offered outside your major department

**Residence Requirements**
For units to be considered in “residence,” you must be registered in courses on the Berkeley campus as a student in the College of Letters & Science. Most students automatically fulfill the residence requirement by attending classes here for four years. In general, there is no need to be concerned about this requirement, unless you go abroad for a semester or year or want to take courses at another institution or through UC Extension during your senior year. In these cases, you should make an appointment to meet an adviser to determine how you can meet the Senior Residence Requirement.

Note: Courses taken through UC Extension do not count toward residence.

**Senior Residence Requirement**
After you become a senior (with 90 semester units earned toward your BA degree), you must complete at least 24 of the remaining 30 units in
residence in at least two semesters. To count as residence, a semester must consist of at least 6 passed units. Intercampus Visitor, EAP, and UC Berkeley-Washington Program (UCDC) units are excluded.

You may use a Berkeley Summer Session to satisfy one semester of the Senior Residence requirement, provided that you successfully complete 6 units of course work in the Summer Session and that you have been enrolled previously in the college.

**Modified Senior Residence Requirement**

Participants in the UC Education Abroad Program (EAP) or the UC Berkeley Washington Program (UCDC) may meet a Modified Senior Residence requirement by completing 24 (excluding EAP) of their final 60 semester units in residence. At least 12 of these 24 units must be completed after you have completed 90 units.

**Upper Division Residence Requirement**

You must complete in residence a minimum of 18 units of upper division courses (excluding EAP units), 12 of which must satisfy the requirements for your major.

**Mission**

The Chemistry major provides training for students planning careers in the chemical sciences and also for those whose interests lie in biology, medicine, earth sciences, secondary education, business, and law. More than half of the total Berkeley undergraduate population will, at some stage of their degree program, take a course from the Department of Chemistry. The curriculum of the department is designed to satisfy the diverse needs of all these students.

Each Chemistry graduate will have completed an integrated, rigorous program which includes foundational course work in chemistry and in-depth course work in chemistry or chemistry-related fields. The ACS-certified degree further emphasizes laboratory experience and the development of professional skills. Advanced coursework and educational activities outside the traditional classroom, such as independent research, provide students the opportunity to conduct individual research projects or participate as a member of a research team. Many undergraduate students also benefit from taking our graduate courses in synthetic and physical chemistry.

At graduation, Chemistry majors will have a set of fundamental competencies that are knowledge-based, performance/skills-based, and effective.

**Learning Goals of the Major**

Graduates will be able to:

1. Master a broad set of chemical knowledge concerning the fundamentals in the basic areas of the discipline (organic, inorganic, analytical, physical, and biological chemistry).
2. Solve problems competently by identifying the essential parts of a problem and formulating a strategy for solving the problem. They will be able to rationally estimate the solution to a problem, apply appropriate techniques to arrive at a solution, test the correctness of the solution, and interpret their results.
3. Use computers in data acquisition and processing and use available software as a tool in data analysis.
4. Employ modern library search tools to locate and retrieve scientific information about a topic, chemical, chemical technique, or an issue relating to chemistry.

**Skills**

Graduates will demonstrate the ability to:

1. Understand the objective of their chemical experiments, properly carry out the experiments, and appropriately record and analyze the results.
2. Use standard laboratory equipment, modern instrumentation, and classical techniques to carry out experiments.
3. Know and follow the proper procedures and regulations for safe handling and use of chemicals.
4. Communicate the concepts and results of their laboratory experiments through effective writing and oral communication skills.

**Effective**

Graduates will be able to:

1. Successfully pursue their career objectives in advanced education in professional and/or graduate schools, in a scientific career in government or industry, in a teaching career in the school systems, or in a related career following graduation.
2. The relationship between the major's core curriculum and student learning outcomes can be seen in the Appendix in Table I.

**Chemistry**

**CHEM 1A General Chemistry 3 Units**

Terms offered: Fall 2018, Summer 2018 8 Week Session, Spring 2018

Stoichiometry of chemical reactions, quantum mechanical description of atoms, the elements and periodic table, chemical bonding, real and ideal gases, thermochemistry, introduction to thermodynamics and equilibrium, acid-base and solubility equilibria, introduction to oxidation-reduction reactions, introduction to chemical kinetics.

**General Chemistry: Read More [+]**

**Rules & Requirements**

**Prerequisites:** High school chemistry recommended

**Credit Restrictions:** Students will receive no credit for Chemistry 1A after completing Chemistry 1AD or 4A. A deficient grade in Chemistry 1A may be removed by taking Chemistry 1AD.<BR/>

**Hours & Format**

Fall and/or spring: 15 weeks - 3 hours of lecture, 1 hour of discussion, and 0 hours of voluntary per week

Summer: 8 weeks - 6 hours of lecture, 2 hours of discussion, and 0 hours of voluntary per week

**Additional Details**

**Subject/Course Level:** Chemistry/Undergraduate

**Grading/Final exam status:** Letter grade. Final exam required.

**General Chemistry:** Read Less [-]
CHEM 1AD General Chemistry (Digital) 3 Units
Terms offered: Spring 2016
An interactive general chemistry course that uses modern digital technology, offered in a smaller classroom setting to facilitate student participation and foster an engaging learning environment. Topics cover the Chemistry 1A curriculum, ranging from quantum mechanics and interactions of atoms and molecules to properties and equilibria of bulk materials. The course involves a blend of classroom lectures and peer learning with substantial web-based assignments and resources including web access to lecture videos. Lecture time is also devoted to ChemQuiz peer discussions and live demos of chemical properties and processes, which students generally find to be illuminating and valuable learning experiences.

Rules & Requirements
Prerequisites: High school chemistry recommended
Credit Restrictions: Students will receive no credit for Chemistry 1AD after completing Chemistry 1A or 4A. A deficient grade in Chemistry 1A may be removed by taking Chemistry 1AD.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture and 1 hour of discussion per week
Summer: 8 weeks - 6 hours of lecture and 2 hours of discussion per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Instructors: Pines, Slack

CHEM 1AL General Chemistry Laboratory 1 Unit
Terms offered: Fall 2018, Summer 2018 8 Week Session, Spring 2018
An experimental approach to chemical sciences with emphasis on developing fundamental, reproducible laboratory technique and a goal of understanding and achieving precision and accuracy in laboratory experiments. Proper use of laboratory equipment and standard wet chemical methods are practiced. Areas of investigations include chemical equilibria, spectroscopy, nanotechnology, green chemistry, and thermochemistry. Concurrent enrollment in 1A is recommended.

Rules & Requirements
Prerequisites: 1A (may be taken concurrently)
Credit Restrictions: Students will receive no credit for 1AL after taking 4A.

Hours & Format
Fall and/or spring: 15 weeks - 1 hour of lecture, 3 hours of laboratory, and 0 hours of voluntary per week
Summer: 8 weeks - 2 hours of lecture, 6 hours of laboratory, and 0 hours of voluntary per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam not required.
CHEM 1B General Chemistry 4 Units
Terms offered: Summer 2018 8 Week Session, Spring 2018, Summer 2017 8 Week Session
Introduction to chemical kinetics, electrochemistry, properties of the states of matter, binary mixtures, thermodynamic efficiency and the direction of chemical change, quantum mechanical description of bonding introduction to spectroscopy. Special topics: Research topics in modern chemistry and biochemistry, chemical engineering.
General Chemistry: Read More [+]

Rules & Requirements

Prerequisites: CHEM 1A and CHEM 1AL with min grades of C-; or CHEM 4A with min grade of C-; or AP CHEM with min score of 4; or CHEM HL IB with min score of 5; or GCE A-Level CHEM with min grade of C

Credit Restrictions: Students will receive no credit for Chemistry 1B after completing Chemistry 4B.

Hours & Format

Fall and/or spring: 15 weeks - 2 hours of lecture, 4 hours of laboratory, and 0 hours of voluntary per week
Summer: 8 weeks - 6 hours of lecture, 8 hours of laboratory, and 0 hours of voluntary per week

Additional Details

Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

General Chemistry: Read Less [-]

CHEM W1A General Chemistry 3 Units
Terms offered: Summer 2013 10 Week Session, Summer 2013 8 Week Session, Summer 2012 8 Week Session
Stoichiometry of chemical reactions, quantum mechanical description of atoms, the elements and periodic table, chemical bonding, real and ideal gases, thermochemistry, introduction to thermodynamics and equilibrium, acid-base and solubility equilibria, introduction to oxidation-reduction reactions, introduction to chemical kinetics. This course is web-based.
General Chemistry: Read More [+]

Rules & Requirements

Prerequisites: High school chemistry is recommended

Credit Restrictions: Students will receive no credit for W1A after taking 1A or 4A. A deficiency in 1A may be removed by taking W1A.

Hours & Format

Fall and/or spring: 15 weeks - 3 hours of web-based lecture and 1 hour of web-based discussion per week
Summer: 8 weeks - 6 hours of web-based lecture and 2 hours of web-based discussion per week

Online: This is an online course.

Additional Details

Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

General Chemistry: Read Less [-]

CHEM 3A Chemical Structure and Reactivity 3 Units
Terms offered: Fall 2018, Summer 2018 8 Week Session, Spring 2018
Introduction to organic chemical structures, bonding, and chemical reactivity. The organic chemistry of alkanes, alkyl halides, alcohols, alkenes, alkynes, and organometallics.
Chemical Structure and Reactivity: Read More [+]

Rules & Requirements

Prerequisites: CHEM 1A with min grade of C-; or AP Chem with min score of 4; or Chem HL IB with min score of 5; or GCE A-Level Chem with min grade of C

Credit Restrictions: Students will receive no credit for CHEM 3A after completing CHEM 12A; a deficient grade in CHEM 12A may be removed by taking CHEM 3A- will restrict credit if completed before Chemistry 3A.

Hours & Format

Fall and/or spring: 15 weeks - 3 hours of lecture per week
Summer: 8 weeks - 6 hours of lecture per week

Additional Details

Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Chemical Structure and Reactivity: Read Less [-]
CHEM 3AL Organic Chemistry Laboratory 2
Units
Terms offered: Fall 2018, Spring 2018, Fall 2017
Introduction to the theory and practice of methods used in the organic
chemistry laboratory. An emphasis is placed on the separation and
purification of organic compounds. Techniques covered will include
extraction, distillation, sublimation, recrystallization, and chromatography.
Detailed discussions and applications of infrared and nuclear magnetic
resonance spectroscopy will be included.
Organic Chemistry Laboratory: Read More [+]

Rules & Requirements
Prerequisites: CHEM 1A and CHEM 1AL with min grades of C-; or
CHEM 4A with min grade of C-; or AP CHEM with min score of 4; or
CHEM HL IB with min score of 5; or GCE A-Level CHEM with min
grade of C.<BR/>Corequisite: CHEM 3A with min grade of C- or coenrollment in CHEM 3A
Credit Restrictions: Students will receive no credit for 3AL after taking
12A.

Hours & Format
Fall and/or spring: 15 weeks - 1 hour of lecture and 4 hours of
laboratory per week
Summer: 8 weeks - 2 hours of lecture and 8 hours of laboratory per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam not required.
Organic Chemistry Laboratory: Read Less [-]

CHEM 3BL Organic Chemistry Laboratory 2
Units
Terms offered: Fall 2018, Summer 2018 8 Week Session, Spring 2018
The synthesis and purification of organic compounds will be explored.
Natural product chemistry will be introduced. Advanced spectroscopic
methods including infrared, ultraviolet, and nuclear magnetic resonance
spectroscopy and mass spectrometry will be used to analyze products
prepared and/or isolated. Qualitative analysis of organic compounds will
be covered.
Organic Chemistry Laboratory: Read More [+]

Rules & Requirements
Prerequisites: CHEM 3AL with min grade of C-<BR/>Corequisite: CHEM 3B with min grade of C- or co-enrollment in CHEM 3B
Credit Restrictions: Students will receive no credit for 3BL after taking
12B.

Hours & Format
Fall and/or spring: 15 weeks - 1 hour of lecture and 4 hours of
laboratory per week
Summer: 8 weeks - 2 hours of lecture and 8 hours of laboratory per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam not required.
Organic Chemistry Laboratory: Read Less [-]

CHEM 3B Chemical Structure and Reactivity
3 Units
Terms offered: Fall 2018, Summer 2018 8 Week Session, Spring 2018
Conjugation, aromatic chemistry, carbonyl compounds, carbohydrates,
amines, carboxylic acids, amino acids, peptides, proteins, and nucleic
acid chemistry. Ultraviolet spectroscopy and mass spectrometry will be
introduced.
Chemical Structure and Reactivity: Read More [+]

Rules & Requirements
Prerequisites: CHEM 3A and CHEM 3AL with min grades of C-
Credit Restrictions: Students will receive no credit for 3B after taking
12B.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week
Summer: 8 weeks - 6 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Chemical Structure and Reactivity: Read Less [-]
CHEM N3AL Organic Chemistry Laboratory 2
Units
Terms offered: Summer 2018 8 Week Session, Summer 2017 8 Week Session, Summer 2015 8 Week Session
Introduction to the theory and practice of methods used in the organic chemistry laboratory. An emphasis is placed on the separation and purification of organic compounds. Techniques covered will include extraction, distillation, sublimation, recrystalization, and chromatography. Detailed discussions and applications of infrared and nuclear magnetic resonance spectroscopy will be included.

Organic Chemistry Laboratory: Read More [+]

Rules & Requirements
Prerequisites: CHEM 1A and CHEM 1AL with min grades of C-; or CHEM 4A with min grade of C-; or AP CHEM with min score of 4; or CHEM HL IB with min score of 5; or GCE A-Level CHEM with min grade of C.<BR/>Co-requisite: CHEM 3A with min grade of C- or co-enrollment in CHEM 3A. Chemistry 4A

Credit Restrictions: Students will receive no credit for N3AL after taking 12A.

Hours & Format
Summer: 8 weeks - 2 hours of web-based lecture and 8 hours of laboratory per week

Online: This is an online course.

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam not required.
Instructor: Pedersen

Organic Chemistry Laboratory: Read Less [-]

CHEM 4A General Chemistry and Quantitative Analysis 4 Units
Terms offered: Fall 2018, Fall 2017, Fall 2016
Series is intended for majors in physical, biological sciences, and engineering. It presents the foundation principles of chemistry, including stoichiometry, ideal and real gases, acid-base and solubility equilibria, oxidation-reduction reactions, thermochemistry, entropy, nuclear chemistry and radioactivity, the atoms and elements, the periodic table, quantum theory, chemical bonding, molecular structure, chemical kinetics, and descriptive chemistry. Examples and applications will be drawn from diverse areas of interest such as atmospheric, environmental, materials, polymer and computational chemistry, and biochemistry. Laboratory emphasizes quantitative work. Equivalent to 1A-1B plus 15 as prerequisite for further courses in chemistry.

General Chemistry and Quantitative Analysis: Read More [+]

Rules & Requirements
Prerequisites: High school chemistry; calculus (may be taken concurrently); high school physics is recommended

Credit Restrictions: Students will receive no credit for 4A after taking 1A. Deficiency in 4A may be removed by successfully completing 1A and 1AL together in the same semester.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture, 4 hours of laboratory, and 0 hours of voluntary per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

General Chemistry and Quantitative Analysis: Read Less [-]
CHEM 4B General Chemistry and Quantitative Analysis 4 Units
Series is intended for majors in physical, biological sciences, and engineering. It presents the foundation principles of chemistry, including stoichiometry, ideal and real gases, acid-base and solubility equilibria, oxidation-reduction reactions, thermochemistry, entropy, nuclear chemistry and radioactivity, the atoms and elements, the periodic table, quantum theory, chemical bonding, molecular structure, chemical kinetics, and descriptive chemistry. Examples and applications will be drawn from diverse areas of interest such as atmospheric, environmental, materials, polymer and computational chemistry, and biochemistry. Laboratory emphasizes quantitative work. Equivalent to 1A-1B plus 15 as prerequisite for further courses in chemistry. General Chemistry and Quantitative Analysis: Read More [+]

Rules & Requirements
Prerequisites: High school chemistry; calculus (may be taken concurrently); high school physics is recommended
Credit Restrictions: Deficiency in 4B may be removed by successfully completing 15.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture, 4 hours of laboratory, and 0 hours of voluntary per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
General Chemistry and Quantitative Analysis: Read Less [-]

CHEM 12A Organic Chemistry 5 Units
Terms offered: Fall 2018, Fall 2017, Fall 1978
A study of all aspects of fundamental organic chemistry, including nomenclature, chemical and physical properties, reactions and syntheses of the major classes of organic compounds. The study includes theoretical aspects, reaction mechanisms, multistep syntheses, and the chemistry of polycyclic and heterocyclic compounds. This course is more extensive and intensive than 3A-3B and includes a greater emphasis on reaction mechanisms and multistep syntheses. 12A (F); 12B (SP)

Organic Chemistry: Read More [+]

Rules & Requirements
Prerequisites: 12A: 1B or 4B with grade of C- or higher; 12B: 12A with grade of C- or higher. For students majoring in chemistry or a closely related field such as chemical engineering or molecular and cell biology
Credit Restrictions: Students will receive no credit for 12A after taking both 3A and 3AL. Deficiency in 12A may be removed by successfully completing 3A and 3AL in the same semester. Students will receive no credit for 12A after taking 112A. Chem 12A is formerly known as Chem 112A.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture, 1 hour of discussion, 5 hours of laboratory, and 0 hours of voluntary per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Formerly known as: Chemistry 112A

Organic Chemistry: Read Less [-]
CHEM 12B Organic Chemistry 5 Units
Terms offered: Spring 2018
A study of all aspects of fundamental organic chemistry, including nomenclature, chemical and physical properties, reactions and syntheses of the major classes of organic compounds. The study includes theoretical aspects, reaction mechanisms, multistep syntheses, and the chemistry of polycyclic and heterocyclic compounds. This course is more extensive and intensive than 3A-3B and includes a greater emphasis on reaction mechanisms and multistep syntheses. 12A (F); 12B (SP)

Rules & Requirements
Prerequisites: 12A: 1B or 4B with grade of C- or higher. 12B: 12A with grade of C- or higher. For students majoring in chemistry or a closely related field such as chemical engineering or molecular and cell biology
Credit Restrictions: Students will receive no credit for 12B after taking both 3B and 3BL. Deficiency in 12B may be removed by successfully completing 3B and 3BL in the same semester. Students will receive no credit for 12B after taking 112B. Chem 12B is formerly known as Chem 112B.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture, 1 hour of discussion, 5 hours of laboratory, and 0 hours of voluntary per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

CHEM 15 Analytical and Bioanalytical Chemistry 3 Units
Terms offered: Fall 2018, Fall 2017, Fall 2016
An introduction to analytical and bioanalytical chemistry including background in statistical analysis of data, acid-base equilibria, electrochemical, spectrometric, and chromatographic methods of analysis and some advanced topics in bioanalytical chemistry such as microfluidics, bioassay techniques, and enzymatic biosensors.

Rules & Requirements
Prerequisites: 1A and 1AL or equivalent
Credit Restrictions: Deficiency in 15 may be removed by successfully completing 4B.

Hours & Format
Fall and/or spring: 15 weeks - 2 hours of lecture and 4 hours of laboratory per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

CHEM 24 Freshman Seminar 1 Unit
Terms offered: Spring 2017, Spring 2016, Spring 2015
The Freshman Seminar Program has been designed to provide new students with the opportunity to explore an intellectual topic with a faculty member in a small seminar setting. Freshman seminars are offered in all campus departments, and topics may vary from department to department and semester to semester. Enrollment limited to 15 freshmen.

Rules & Requirements
Repeat rules: Course may be repeated for credit when topic changes. Course may be repeated for credit when topic changes.

Hours & Format
Fall and/or spring: 15 weeks - 1 hour of seminar per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: The grading option will be decided by the instructor when the class is offered. Final exam required.

CHEM 49 Supplementary Work in Lower Division Chemistry 1 - 4 Units
Terms offered: Fall 2017, Fall 2016, Spring 2016
Students with partial credit in lower division chemistry courses may, with consent of instructor, complete the credit under this heading.

Rules & Requirements
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 1-4 hours of independent study per week
Summer:
6 weeks - 1-6 hours of independent study per week
8 weeks - 1-4 hours of independent study per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam not required.
CHEM 96 Introduction to Research and Study in the College of Chemistry 1 Unit
Terms offered: Fall 2017, Fall 2016, Fall 2015
Introduces freshmen to research activities and programs of study in the College of Chemistry. Includes lectures by faculty, an introduction to college library and computer facilities, the opportunity to meet alumni and advanced undergraduates in an informal atmosphere, and discussion of college and campus resources.
Introduction to Research and Study in the College of Chemistry: Read More [+]

Rules & Requirements
Prerequisites: Freshman standing in the College of Chemistry, or consent of instructor
Credit Restrictions: Students will receive no credit for Chemistry 96 after taking Chemistry C96 or Chemical and Biomolecular Engineering C96.

Hours & Format
Fall and/or spring: 15 weeks - 1 hour of seminar per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Offered for pass/not pass grade only. Final exam required.

Introduction to Research and Study in the College of Chemistry: Read Less [-]

CHEM 98 Supervised Group Study 1 - 4 Units
Terms offered: Fall 2018, Fall 2017, Spring 2017
Group study of selected topics.
Supervised Group Study: Read More [+]

Rules & Requirements
Prerequisites: Consent of instructor
Credit Restrictions: Enrollment is restricted; see the Introduction to Courses and Curricula section of this catalog.

Hours & Format
Fall and/or spring: 15 weeks - 1-4 hours of directed group study per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Offered for pass/not pass grade only. Final exam not required.

Supervised Group Study: Read Less [-]

CHEM 98W Directed Group Study 1 Unit
Terms offered: Spring 2013, Fall 2012, Spring 2009
Topics vary with instructor. Enrollment restrictions apply.
Directed Group Study: Read More [+]

Rules & Requirements
Credit Restrictions: Enrollment is restricted; see the Introduction to Courses and Curricula section of this catalog.
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 1 hour of directed group study per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Offered for pass/not pass grade only. Final exam not required.

Directed Group Study: Read Less [-]

CHEM 100 Communicating Chemistry 2 Units
Terms offered: Spring 2011, Spring 2010, Spring 2009
For undergraduate and graduate students interested in improving their ability to communicate their scientific knowledge by teaching chemistry in elementary schools. The course will combine instruction in inquiry-based chemistry teaching methods and learning pedagogy with 10 weeks of supervised teaching experience in a local school classroom. Thus, students will practice communicating scientific knowledge and receive mentoring on how to improve their presentations. Approximately three hours per week, including time spent in school classrooms.
Communicating Chemistry: Read More [+]

Rules & Requirements
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 2 hours of lecture and 1 hour of fieldwork per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Formerly known as: 20
CHEM 103 Inorganic Chemistry in Living Systems 3 Units
Terms offered: Fall 2018, Fall 2017, Fall 2016
The basic principles of metal ions and coordination chemistry applied to the study of biological systems.
Inorganic Chemistry in Living Systems: Read More

Rules & Requirements
Prerequisites: Chemistry 3A or 112A. Chemistry majors can only count 2 of the 3 units towards their Allied Subject requirement

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

CHEM 104A Advanced Inorganic Chemistry 3 Units
Terms offered: Fall 2018, Fall 2017, Fall 2016
The chemistry of metals and nonmetals including the application of physical chemical principles.
Advanced Inorganic Chemistry: Read More

Rules & Requirements
Prerequisites: 1B, 4B, or 3A; 104A is prerequisite to 104B
Credit Restrictions: 104A: No restrictions; 104B: Students will receive two units of credit after taking 103.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week
Summer: 8 weeks - 6 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

CHEM 104B Advanced Inorganic Chemistry 3 Units
The chemistry of metals and nonmetals including the application of physical chemical principles.
Advanced Inorganic Chemistry: Read More

Rules & Requirements
Prerequisites: 104A or consent of instructor. Chemical Biology majors can only count 2 of the 3 units towards their Allied Subject requirement for 104B after taking 103

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week
Summer: 8 weeks - 6 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

CHEM 105 Instrumental Methods in Analytical Chemistry 4 Units
Principles, instrumentation and analytical applications of atomic spectroscopies, mass spectrometry, separations, electrochemistry and micro-characterization. Discussion of instrument design and capabilities as well as real-world problem solving with an emphasis on bioanalytical, environmental, and forensic applications. Hands-on laboratory work using modern instrumentation, emphasizing independent projects involving real-life samples and problem solving.
Inorganic Chemistry in Living Systems: Read More

Rules & Requirements
Prerequisites: 4B; or 1B and 15; or 1B and a UC GPA of 3.3 or higher

Hours & Format
Fall and/or spring: 15 weeks - 2 hours of lecture and 8 hours of laboratory per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
CHEM 108 Inorganic Synthesis and Reactions 4 Units
Terms offered: Fall 2018, Fall 2017, Fall 2016
The preparation of inorganic compounds using vacuum line, air-and
moisture-exclusion, electrochemical, high-pressure, and other synthetic
techniques. Kinetic and mechanistic studies of inorganic compounds.
Inorganic Synthesis and Reactions: Read More [+]

Rules & Requirements
Prerequisites: 4B or 15; 104B with grade of C- or higher, or 103

Hours & Format
Fall and/or spring: 15 weeks - 2 hours of lecture and 8 hours of
laboratory per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Inorganic Synthesis and Reactions: Read Less [-]

CHEM C110L General Biochemistry and Molecular Biology Laboratory 4 Units
Terms offered: Fall 2018, Spring 2018, Fall 2017
Experimental techniques of biochemistry and molecular biology, designed
to accompany the lectures in Molecular and Cell Biology 100B and 110.
General Biochemistry and Molecular Biology Laboratory: Read More [+]

Rules & Requirements
Prerequisites: 110 (may be taken concurrently)

Hours & Format
Fall and/or spring: 15 weeks - 2-2 hours of lecture and 6-8 hours of
laboratory per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam not required.
Also listed as: MCELLBI C110L

General Biochemistry and Molecular Biology Laboratory: Read Less [-]

CHEM 113 Advanced Mechanistic Organic Chemistry 3 Units
Terms offered: Fall 2018, Fall 2016, Fall 2015
Advanced topics in mechanistic and physical organic chemistry typically
including kinetics, reactive intermediates, substitution reactions, linear
free energy relationships, orbital interactions and orbital symmetry control
of reactions, isotope effects, and photochemistry.
Advanced Mechanistic Organic Chemistry: Read More [+]

Rules & Requirements
Prerequisites: 3B or 112B with a minimum grade of B- or consent of
instructor

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Advanced Mechanistic Organic Chemistry: Read Less [-]

CHEM 114 Advanced Synthetic Organic Chemistry 3 Units
Terms offered: Spring 2018, Spring 2017, Spring 2015
Advanced topics in synthetic organic chemistry with a focus on
selectivity. Topics include reductions, oxidations, enolate chemistry
and the aldol reaction, reactions of non-stabilized anions, olefination
reactions, pericyclic reactions and application to the synthesis of complex
structures.
Advanced Synthetic Organic Chemistry: Read More [+]

Rules & Requirements
Prerequisites: 3B or 112B with a minimum grade of B- or consent of
instructor

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Advanced Synthetic Organic Chemistry: Read Less [-]
CHEM 115 Organic Chemistry--Advanced Laboratory Methods 4 Units
Terms offered: Fall 2018, Summer 2018 First 6 Week Session, Spring 2018
Advanced synthetic methods, chemical and spectroscopic structural methods, designed as a preparation for experimental research.
Organic Chemistry--Advanced Laboratory Methods: Read More [+]
Rules & Requirements
Prerequisites: 112B with a grade of C- or higher
Hours & Format
Fall and/or spring: 15 weeks - 1 hour of lecture and 11 hours of laboratory per week
Summer:
6 weeks - 2.5 hours of lecture and 27.5 hours of laboratory per week
8 weeks - 2 hours of lecture and 20.5 hours of laboratory per week
Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Alternative to final exam.
Organic Chemistry--Advanced Laboratory Methods: Read Less [-]

CHEM 120B Physical Chemistry 3 Units
Terms offered: Fall 2018, Fall 2017, Fall 2016
Statistical mechanics, thermodynamics, equilibrium and applications to chemical systems: states of matter, solutions and solvation, chemical kinetics, molecular dynamics, and molecular transport.
Physical Chemistry: Read More [+]
Rules & Requirements
Prerequisites: 4B or equivalent; Mathematics 53; Mathematics 54 (may be taken concurrently); Physics 7B or 8B
Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week
Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Physical Chemistry: Read Less [-]

CHEM 120A Physical Chemistry 3 Units
Terms offered: Fall 2018, Spring 2018, Fall 2017
Kinetic, potential, and total energy of particles and forces between them; principles of quantum theory, including one-electron and many-electron atoms and its applications to chemical bonding, intermolecular interactions, and elementary spectroscopy.
Physical Chemistry: Read More [+]
Rules & Requirements
Prerequisites: 4B or equivalent; Physics 7B or 8B; Mathematics 53; Mathematics 54 or consent of instructor
Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture and 0 hours of voluntary per week
Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Physical Chemistry: Read Less [-]

CHEM 122 Quantum Mechanics and Spectroscopy 3 Units
Terms offered: Fall 2017, Spring 2017, Spring 2015
Postulates and methods of quantum mechanics and group theory applied to molecular structure and spectra.
Quantum Mechanics and Spectroscopy: Read More [+]
Rules & Requirements
Prerequisites: 120A
Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week
Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Quantum Mechanics and Spectroscopy: Read Less [-]
CHEM 125 Physical Chemistry Laboratory 3 Units
Terms offered: Fall 2018, Spring 2018, Fall 2017
Experiments in thermodynamics, kinetics, molecular structure, and general physical chemistry.
Physical Chemistry Laboratory: Read More [+]

Rules & Requirements

Prerequisites: Two of the following: 120A, 120B, C130, or 130B with grades of C- or higher (one of which may be taken concurrently)

Credit Restrictions: Deficiency in 125 may be removed by successfully completing C182. Consent of instructor is required to enroll in 125 after completing C182 or EPS C182.

Hours & Format

Fall and/or spring: 15 weeks - 1 hour of lecture and 5 hours of laboratory per week

Additional Details

Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Physical Chemistry Laboratory: Read Less [-]

CHEM C130 Biophysical Chemistry: Physical Principles and the Molecules of Life 4 Units
Terms offered: Fall 2018, Spring 2018, Fall 2017
Thermodynamic and kinetic concepts applied to understanding the chemistry and structure of biomolecules (proteins, DNA, and RNA). Molecular distributions, reaction kinetics, enzyme kinetics. Bioenergetics, energy transduction, and motor proteins. Electrochemical potential, membranes, and ion channels.

Biophysical Chemistry: Physical Principles and the Molecules of Life: Read More [+]

Rules & Requirements

Prerequisites: Chemistry 3A or 112A, Mathematics 1A, Biology 1A and 1AL; Chemistry 3B or 112B recommended

Hours & Format

Fall and/or spring: 15 weeks - 3 hours of lecture and 1 hour of discussion per week
Summer: 8 weeks - 5.5 hours of lecture and 2 hours of discussion per week

Additional Details

Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Biophysical Chemistry: Physical Principles and the Molecules of Life: Read Less [-]

CHEM 130B Biophysical Chemistry 3 Units
The weekly one-hour discussion is for problem solving and the application of calculus in physical chemistry. Molecular structure, intermolecular forces and interactions, biomolecular spectroscopy, high-resolution structure determinations.

Biophysical Chemistry: Read More [+]

Rules & Requirements

Prerequisites: Chemistry C130 or Molecular and Cell Biology C100A, or consent of instructor. Chemistry and Chemical Biology majors can only count 2 of the 3 units towards their Allied Subject requirement

Hours & Format

Fall and/or spring: 15 weeks - 2 hours of lecture and 1 hour of discussion per week

Additional Details

Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Biophysical Chemistry: Read Less [-]

CHEM 135 Chemical Biology 3 Units
Terms offered: Fall 2018, Fall 2017, Fall 2016
One-semester introduction to biochemistry, aimed toward chemistry and chemical biology majors.

Chemical Biology: Read More [+]

Rules & Requirements

Prerequisites: 3B or 112B; Biology 1A; or consent of instructor

Credit Restrictions: Students will receive no credit for 135 after taking Molecular and Cell Biology 100B or 102.

Hours & Format

Fall and/or spring: 15 weeks - 3 hours of lecture per week

Additional Details

Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Chemical Biology: Read Less [-]
CHEM 146 Radiochemical Methods in Nuclear Technology and Forensics 3 Units
Experimental illustrations of the interrelation between chemical and nuclear science and technology and nuclear forensics; radioactive decay and counting techniques; nuclear spectroscopy; fundamental radiochemical techniques; radiochemical separations techniques; tracers; activation analysis; forensic applications of radiochemistry; fusion, fission and nuclear reactors.
Radiochemical Methods in Nuclear Technology and Forensics: Read More [+]

Objectives Outcomes
Course Objectives: Familiarize students with principles of nuclear and radiochemistry and its many important applications in our daily lives; provide hands-on training.
Student Learning Outcomes: A solid understanding of nuclear and radiochemistry; proficiency in safe handling of radioactive materials in the laboratory, and appreciation for the wide application of radiochemical techniques in chemistry, nuclear technology, and nuclear forensics.

Rules & Requirements
Prerequisites: 4B or 15; 143 is recommended

Hours & Format
Fall and/or spring: 15 weeks - 1.5 hours of lecture and 4.5 hours of laboratory per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Instructor: Nitsche

Radiochemical Methods in Nuclear Technology and Forensics: Read Less [-]

CHEM 143 Nuclear Chemistry 2 Units
Terms offered: Fall 2018, Fall 2017, Fall 2016
Radioactivity, fission, nuclear models and reactions, nuclear processes in nature. Computer methods will be introduced.
Nuclear Chemistry: Read More [+]

Rules & Requirements
Prerequisites: Physics 7B or equivalent

Hours & Format
Fall and/or spring: 15 weeks - 2 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Instructor: Nitsche

Radiochemical Methods in Nuclear Technology and Forensics: Read Less [-]

CHEM C138 The Berkeley Lectures on Energy: Energy from Biomass 3 Units
Terms offered: Fall 2015, Fall 2014, Fall 2013
After an introduction to the different aspects of our global energy consumption, the course will focus on the role of biomass. The course will illustrate how the global scale of energy guides the biomass research. Emphasis will be placed on the integration of the biological aspects (crop selection, harvesting, storage and distribution, and chemical composition of biomass) with the chemical aspects to convert biomass to energy. The course aims to engage students in state-of-the-art research.
The Berkeley Lectures on Energy: Energy from Biomass: Read More [+]

Rules & Requirements
Prerequisites: Chemistry 1B or Chemistry 4B, Mathematics 1B, Biology 1A

Repeat rules: Course may be repeated for credit under special circumstances: Repeatable when topic changes with consent of instructor. Repeatable when topic changes with consent of instructor.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Instructors: Bell, Blanch, Clark, Smit, C. Somerville
Also listed as: BIO ENG C181/CHM ENG C195A/PLANTBI C124

The Berkeley Lectures on Energy: Energy from Biomass: Read Less [-]
CHEM 149 Supplementary Work in Upper Division Chemistry 1 - 4 Units
Terms offered: Spring 2016, Spring 2015, Spring 2014
Students with partial credit in upper division chemistry courses may, with consent of instructor, complete the credit under this heading.
Supplementary Work in Upper Division Chemistry: Read More [+]

Rules & Requirements
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 1-4 hours of independent study per week
Summer:
6 weeks - 2.5-10 hours of independent study per week
8 weeks - 1.5-7.5 hours of independent study per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

CHEM C150 Introduction to Materials Chemistry 3 Units
The application of basic chemical principles to problems in materials discovery, design, and characterization will be discussed. Topics covered will include inorganic solids, nanoscale materials, polymers, and biological materials, with specific focus on the ways in which atomic-level interactions dictate the bulk properties of matter.
Introduction to Materials Chemistry: Read More [+]

Rules & Requirements
Prerequisites: 104A; 104B is recommended

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Also listed as: MAT SCI C150

Introduction to Materials Chemistry: Read Less [-]

CHEM C170L Biochemical Engineering Laboratory 3 Units
Terms offered: Fall 2018, Spring 2018, Spring 2014, Spring 2013
Laboratory techniques for the cultivation of microorganisms in batch and continuous reactions. Enzymatic conversion processes. Recovery of biological products.
Biochemical Engineering Laboratory: Read More [+]

Rules & Requirements
Prerequisites: Chemical Engineering 170A (may be taken concurrently) or consent of instructor

Hours & Format
Fall and/or spring: 15 weeks - 7 hours of laboratory and 1 hour of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Also listed as: CHM ENG C170L

Biochemical Engineering Laboratory: Read Less [-]

CHEM C178 Polymer Science and Technology 3 Units
Terms offered: Spring 2018, Fall 2016, Spring 2016, Spring 2015
An interdisciplinary course on the synthesis, characterization, and properties of polymer materials. Emphasis on the molecular origin of properties of polymeric materials and technological applications. Topics include single molecule properties, polymer mixtures and solutions, melts, glasses, elastomers, and crystals. Experiments in polymer synthesis, characterization, and physical properties.
Polymer Science and Technology: Read More [+]

Rules & Requirements
Prerequisites: Junior standing

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture and 3 hours of laboratory per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.

Instructor: Segalman

Also listed as: CHM ENG C178

Polymer Science and Technology: Read Less [-]
CHEM C182 Atmospheric Chemistry and Physics Laboratory 3 Units
Fluid dynamics, radiative transfer, and the kinetics, spectroscopy, and measurement of atmospherically relevant species are explored through laboratory experiments, numerical simulations, and field observations. Atmospheric Chemistry and Physics Laboratory: Read More [+]

Rules & Requirements
Prerequisites: Earth and Planetary Science 50 and 102 with grades of C- or higher (one of which may be taken concurrently) or two of the following: Chemistry 120A, 120B, C130, or 130B with grades of C- or higher (one of which may be taken concurrently)
Credit Restrictions: Deficiency in C182 may be removed by successfully completing 125. Consent of instructor is required to enroll in C182 after completing 125.

Hours & Format
Fall and/or spring: 15 weeks - 1.5 hours of lecture and 5 hours of laboratory per week
Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Also listed as: EPS C182
Atmospheric Chemistry and Physics Laboratory: Read Less [-]

CHEM C191 Quantum Information Science and Technology 3 Units
Terms offered: Spring 2017, Fall 2014, Spring 2012
This multidisciplinary course provides an introduction to fundamental conceptual aspects of quantum mechanics from a computational and informational theoretic perspective, as well as physical implementations and technological applications of quantum information science. Basic sections of quantum algorithms, complexity, and cryptography, will be touched upon, as well as pertinent physical realizations from nanoscale science and engineering.
Quantum Information Science and Technology: Read More [+]

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week
Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Instructors: Crommie, Vazirani, Whaley
Also listed as: COMPSCI C191/PHYSICS C191
Quantum Information Science and Technology: Read Less [-]

CHEM 192 Individual Study for Advanced Undergraduates 1 - 3 Units
Terms offered: Spring 2016, Fall 2015, Spring 2015
All properly qualified students who wish to pursue a problem of their own choice, through reading or nonlaboratory study, may do so if their proposed project is acceptable to the member of the staff with whom they wish to work.
Individual Study for Advanced Undergraduates: Read More [+]

Rules & Requirements
Prerequisites: Consent of instructor and adviser
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 0 hours of independent study per week
Summer:
6 weeks - 1-5 hours of independent study per week
8 weeks - 1-4 hours of independent study per week
Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam not required.
Individual Study for Advanced Undergraduates: Read Less [-]

CHEM H193 Senior Honors Thesis 3 Units
Terms offered: Spring 2016, Fall 2015, Spring 2015
A senior honors thesis is written in consultation with the student’s faculty research advisor. This is a required course for students wishing to graduate with honors in Chemistry or Chemical Biology.
Senior Honors Thesis: Read More [+]

Rules & Requirements
Prerequisites: Senior standing, approval of faculty research advisor, overall GPA of 3.4 or higher at Berkeley
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 9-0 hours of independent study per week
Summer: 8 weeks - 16.5 hours of independent study per week
Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Alternative to final exam.
Senior Honors Thesis: Read Less [-]
CHEM H194 Research for Advanced Undergraduates 2 - 4 Units
Terms offered: Fall 2016, Summer 2016 8 Week Session, Spring 2016
Students may pursue original research under the direction of one of the members of the staff.
Research for Advanced Undergraduates: Read More [+]

Rules & Requirements
Prerequisites: Minimum GPA of 3.4 overall at Berkeley and consent of instructor and adviser
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 0 hours of independent study per week
Summer:
0 weeks - 2-4 hours of independent study per week
6 weeks - 2-4 hours of independent study per week
8 weeks - 2-4 hours of independent study per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam not required.
Research for Advanced Undergraduates: Read Less [-]

CHEM 195 Special Topics 3 Units
Terms offered: Fall 2018, Spring 2018, Spring 2016
Special topics will be offered from time to time. Examples are: photochemical air pollution, computers in chemistry.
Special Topics: Read More [+]

Rules & Requirements
Prerequisites: Consent of instructor
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of lecture per week
Summer: 10 weeks - 4.5 hours of lecture per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Special Topics: Read Less [-]

CHEM 196 Special Laboratory Study 2 - 4 Units
Terms offered: Spring 2016, Fall 2015, Summer 2015 8 Week Session
Special laboratory work for advanced undergraduates.
Special Laboratory Study: Read More [+]

Rules & Requirements
Prerequisites: Consent of instructor and adviser
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 1-1 hours of independent study per week
Summer:
6 weeks - 2.5-2.5 hours of independent study per week
8 weeks - 2-2 hours of independent study per week
10 weeks - 1.5-1.5 hours of independent study per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Letter grade. Final exam required.
Special Laboratory Study: Read Less [-]

CHEM 197 Field Study in Chemistry 1 - 4 Units
Terms offered: Summer 2016 8 Week Session, Spring 2016, Summer 2015 8 Week Session
Supervised experience in off-campus organizations relevant to specific aspects and applications of chemistry. Written report required at the end of the term. Course does not satisfy unit or residence requirements for the bachelor's degree.
Field Study in Chemistry: Read More [+]

Rules & Requirements
Prerequisites: Upper division standing and consent of instructor
Credit Restrictions: Enrollment is restricted; see the Introduction to Courses and Curricula section of this catalog.
Repeat rules: Course may be repeated for credit without restriction.

Hours & Format
Fall and/or spring: 15 weeks - 3 hours of fieldwork per week
Summer: 8 weeks - 6 hours of fieldwork per week

Additional Details
Subject/Course Level: Chemistry/Undergraduate
Grading/Final exam status: Offered for pass/not pass grade only. Final exam not required.
Field Study in Chemistry: Read Less [-]
CHEM 198 Directed Group Study 1 - 4 Units
Terms offered: Fall 2018, Summer 2018 Second 6 Week Session, Fall 2017
Group study of selected topics.
Directed Group Study: Read More [+]

Rules & Requirements

Prerequisites: Completion of 60 units of undergraduate study and in good standing

Credit Restrictions: Enrollment is restricted; see the Introduction to Courses and Curricula section of this catalog.

Repeat rules: Course may be repeated for credit without restriction.

Hours & Format

Fall and/or spring: 15 weeks - 1-4 hours of directed group study per week

Additional Details

Subject/Course Level: Chemistry/Undergraduate

Grading/Final exam status: Offered for pass/not pass grade only. Final exam not required.

Directed Group Study: Read Less [-]

CHEM 199 Supervised Independent Study and Research 1 - 4 Units
Terms offered: Spring 2016, Fall 2015, Spring 2015
Enrollment is restricted by regulations listed in the .
Supervised Independent Study and Research: Read More [+]

Rules & Requirements

Repeat rules: Course may be repeated for credit without restriction.

Hours & Format

Fall and/or spring: 15 weeks - 0 hours of independent study per week

Summer:
6 weeks - 1-5 hours of independent study per week
8 weeks - 1-4 hours of independent study per week

Additional Details

Subject/Course Level: Chemistry/Undergraduate

Grading/Final exam status: Offered for pass/not pass grade only. Final exam not required.

Supervised Independent Study and Research: Read Less [-]